# **Acid Base Disorders**

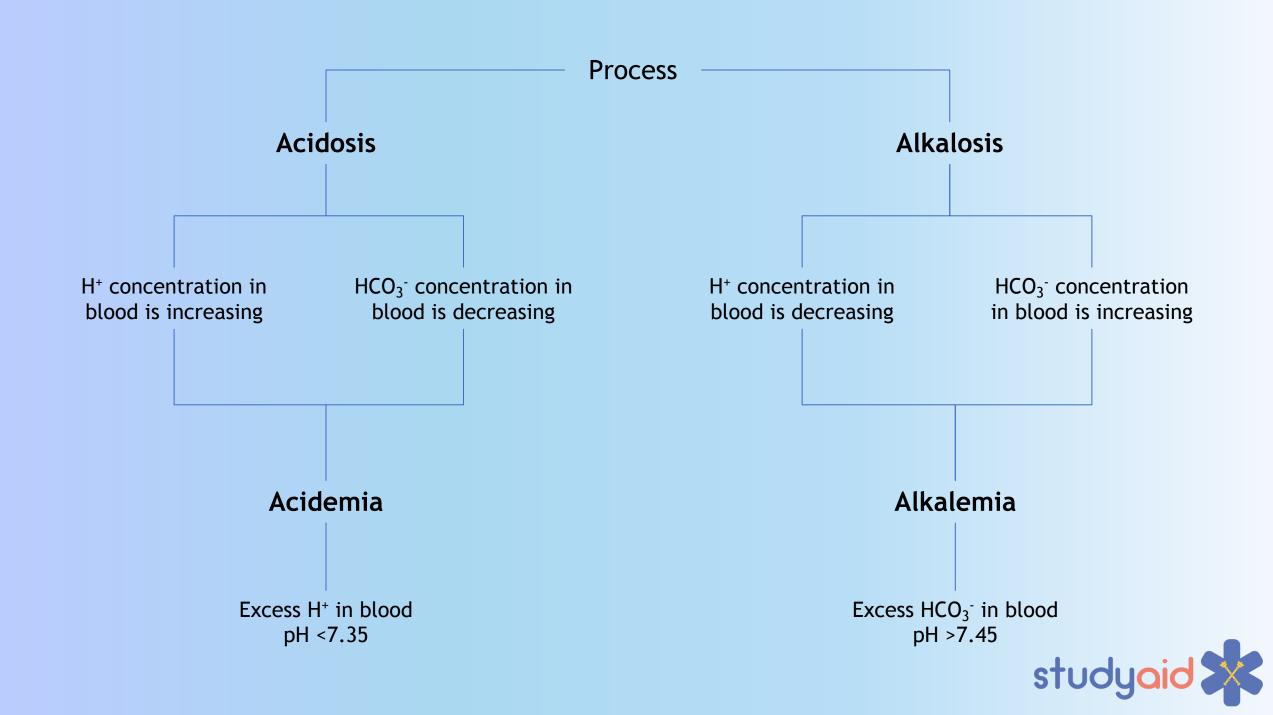
Nora Sønstebø



## Arterial blood gas (ABG)

- pH 7.35-7.45
- pCO<sub>2</sub> 35-45 mmHg
- HCO<sub>3</sub>-22-26 mEq/L
- pO<sub>2</sub> 75-100 mmHg

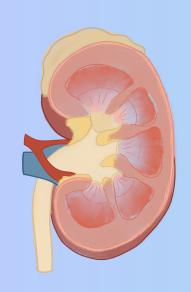




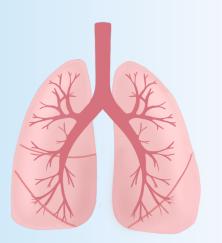


VS

 $CO_2$ 



Base Kidney Metabolic Days Acid
Lungs
Respiratory
Minutes

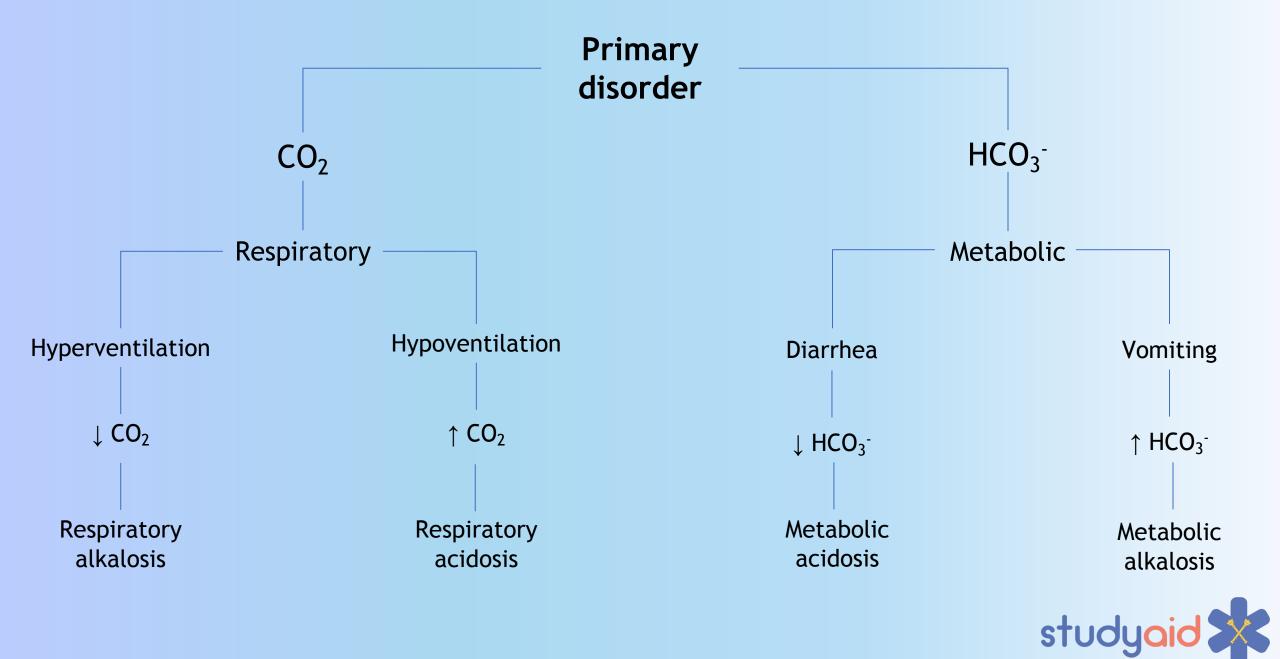




## 3-step ABG interpretation

- 1. Identify if it is acidosis or alkalosis (pH)
- 2. Identify if it is **respiratory** or **metabolic** (CO<sub>2</sub>/HCO<sub>3</sub><sup>-</sup>)
- 3. Identify it if is compensated (partially or fully) or uncompensated





pH 7.20

pCO<sub>2</sub> 40

HCO<sub>3</sub>- 15

pH 7.35-7.45

pCO<sub>2</sub> 35-45 mmHg



pH 7.50

pCO<sub>2</sub> 31

HCO<sub>3</sub>- 24

pH 7.35-7.45

pCO<sub>2</sub> 35-45 mmHg



pH 7.31

pCO<sub>2</sub> 28

HCO<sub>3</sub>- 18

pH 7.35-7.45

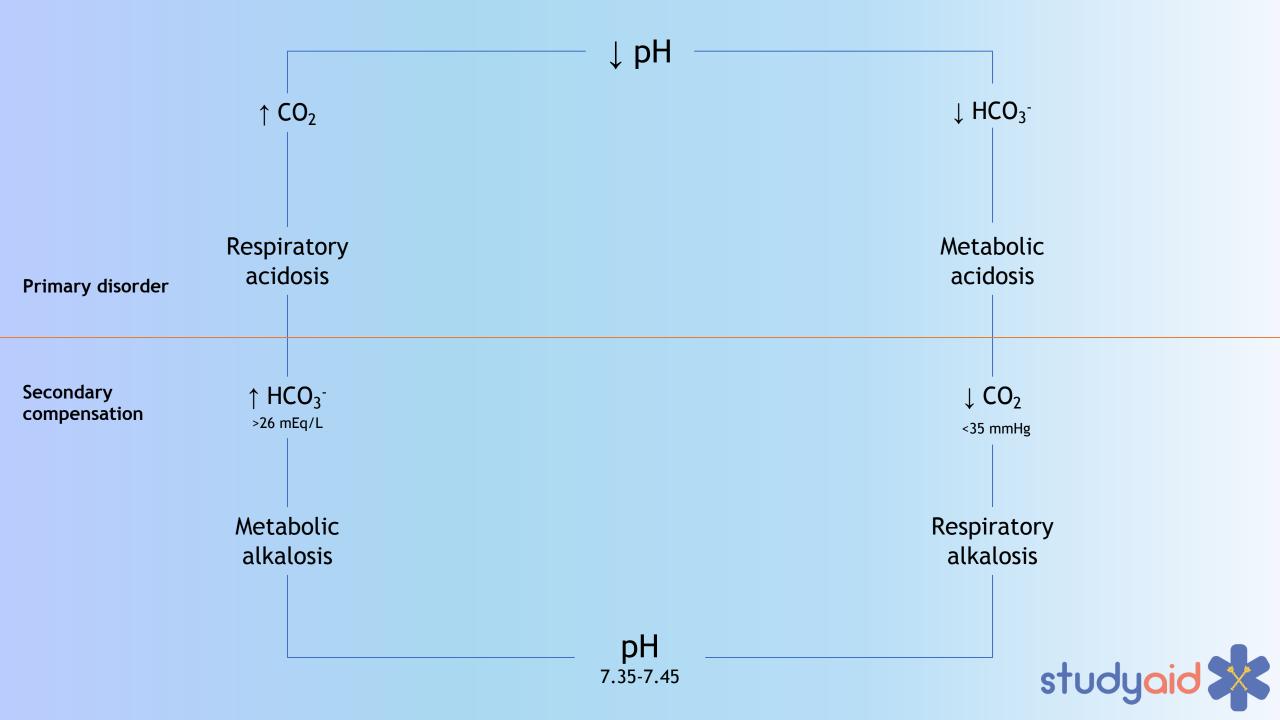
pCO<sub>2</sub> 35-45 mmHg

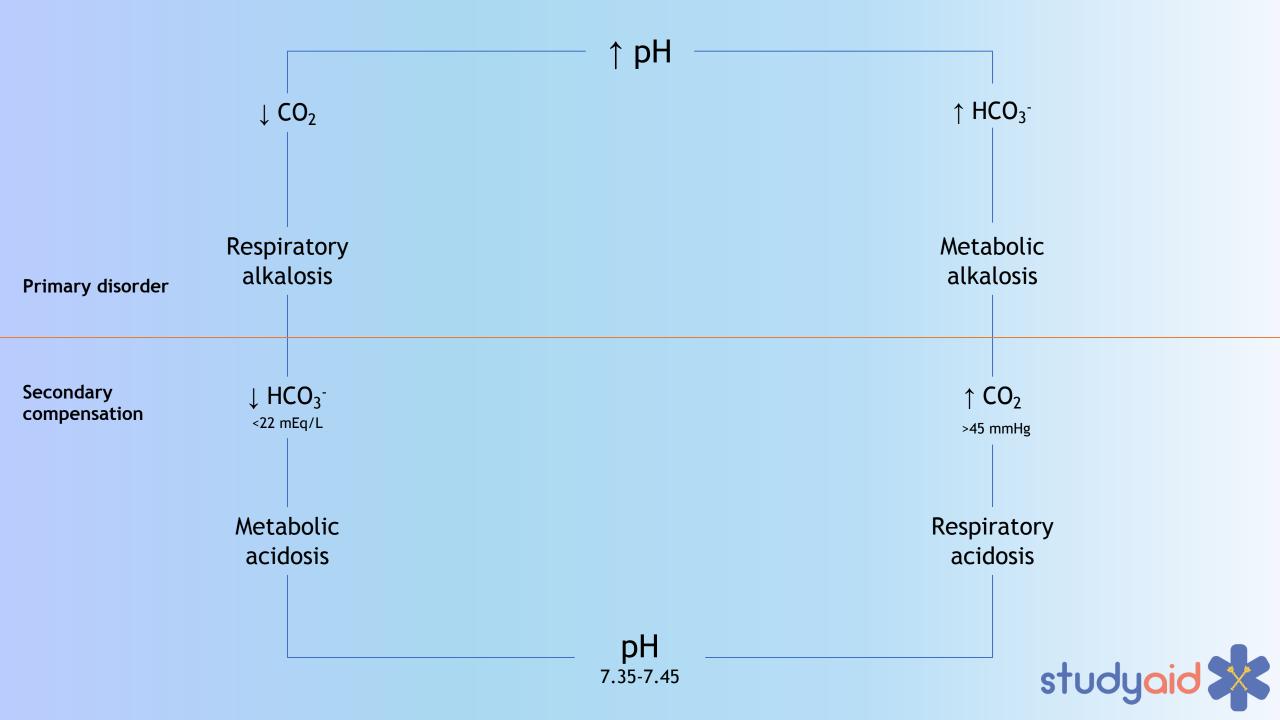


## 3-step ABG interpretation

- 1. Identify if it is acidosis or alkalosis (pH)
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pH 7.31

pCO<sub>2</sub> 28

HCO<sub>3</sub>- 18

pH 7.35-7.45

pCO<sub>2</sub> 35-45 mmHg



pH 7.5

pCO<sub>2</sub> 51

HCO<sub>3</sub>- 28

pH 7.35-7.45

pCO<sub>2</sub> 35-45 mmHg



pH 7.35

pCO<sub>2</sub> 49

HCO<sub>3</sub>- 30

pH 7.35-7.45

pCO<sub>2</sub> 35-45 mmHg



### Full compensation:

- pH is normal
- CO<sub>2</sub> and HCO<sub>3</sub>- levels are abnormal

### Acidic side / alkaline side

- 7.35, 7.36, 7.37, 7.38, 7.39 Acidic side
- 7.41, 7.42, 7.43, 7.44, 7.45 Alkaline side



pH 7.35

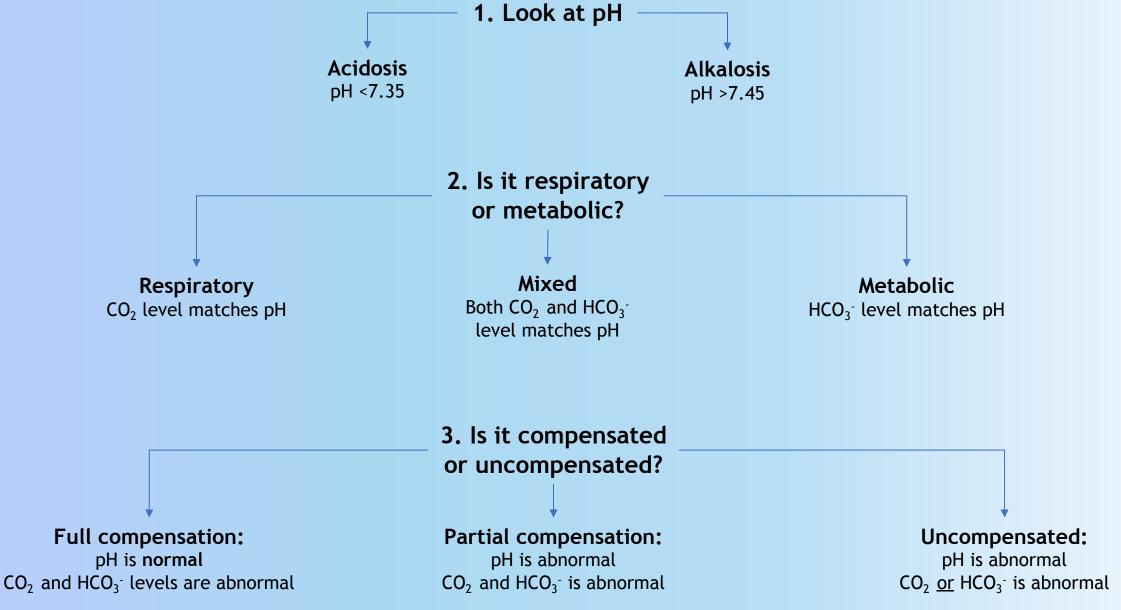
pCO<sub>2</sub> 49

HCO<sub>3</sub>- 30

pH 7.35-7.45

pCO<sub>2</sub> 35-45 mmHg







George W. Bush is diagnosed with peptic ulcer disease, which he take antacids for. He visits you, his private doctor, and presents with confusion and hypoventilation. You take an ABG sample, which reveals:

pH 7.42

pCO<sub>2</sub> 50

HCO<sub>3</sub>- 33

How would you classify the acid base status?





pH 7.35-7.45

pCO<sub>2</sub> 35-45 mmHg



Frank Løke is coming down from the summit of Mount Everest. At an altitude of 8,400 meters, he has a blood gas drawn while breathing ambient air as part of a research project.

#### Arterial blood gas reveals:

pH 7.55

pCO<sub>2</sub> 12

 $HCO_{3}^{-}$  10.5



How would you classify the acid base status?



7.35 caused by \$\psi\$ HCO <sub>3</sub> - ncentration in blood  7.45 caused by \$\psi\$ HCO <sub>3</sub> - ncentration in blood	Hyperventilation  ↓ CO₂  Compensation occurs within minutes  Hypoventilation  ↑ CO₂  Compensation occurs within minutes	$\downarrow pH$ $\downarrow HCO_{3}^{-}$ $\downarrow CO_{2} \text{ (compensation)}$ $\uparrow pH$ $\uparrow HCO_{3}^{-}$ $\uparrow CO_{2} \text{ (compensation)}$
ncentration in blood  7.45 caused by $\uparrow$ HCO <sub>3</sub>	Compensation occurs within minutes  Hypoventilation $\uparrow \text{CO}_2$ Compensation occurs within	$\downarrow$ CO <sub>2</sub> (compensation) $\uparrow$ pH $\uparrow$ HCO <sub>3</sub> -
7.45 caused by ↑ HCO <sub>3</sub> -	minutes  Hypoventilation  ↑ CO <sub>2</sub> Compensation occurs within	↑ pH ↑ HCO <sub>3</sub> -
	Hypoventilation  ↑ CO <sub>2</sub> Compensation occurs within	↑ HCO <sub>3</sub> -
	↑ CO <sub>2</sub> Compensation occurs within	↑ HCO <sub>3</sub> -
	Compensation occurs within	
ncentration in blood		↑ CO <sub>2</sub> (compensation)
	minutes	
	↑ renal reabsorption of HCO <sub>3</sub> -	↓ pH
<7.35 caused by ↑ CO <sub>2</sub>	Compensation occurs within	↑ CO <sub>2</sub>
concentration in blood	hours to days	↑ HCO <sub>3</sub> - (compensation)
Respiratory alkalosis  pH >7.45 caused by ↓ CO₂  concentration in blood	↓ renal reabsorption of HCO <sub>3</sub> -	↑ pH
	Compensation occurs within	↓ CO <sub>2</sub>
	hours to days	↓ HCO <sub>3</sub> - (compensation)
r -	ncentration in blood  7.45 caused by $\downarrow$ CO <sub>2</sub>	hours to days $\downarrow \text{ renal reabsorption of HCO}_3^-$ 7.45 caused by $\downarrow \text{CO}_2$ Compensation occurs within

# https://abg.ninja/abg









2 You can participate

